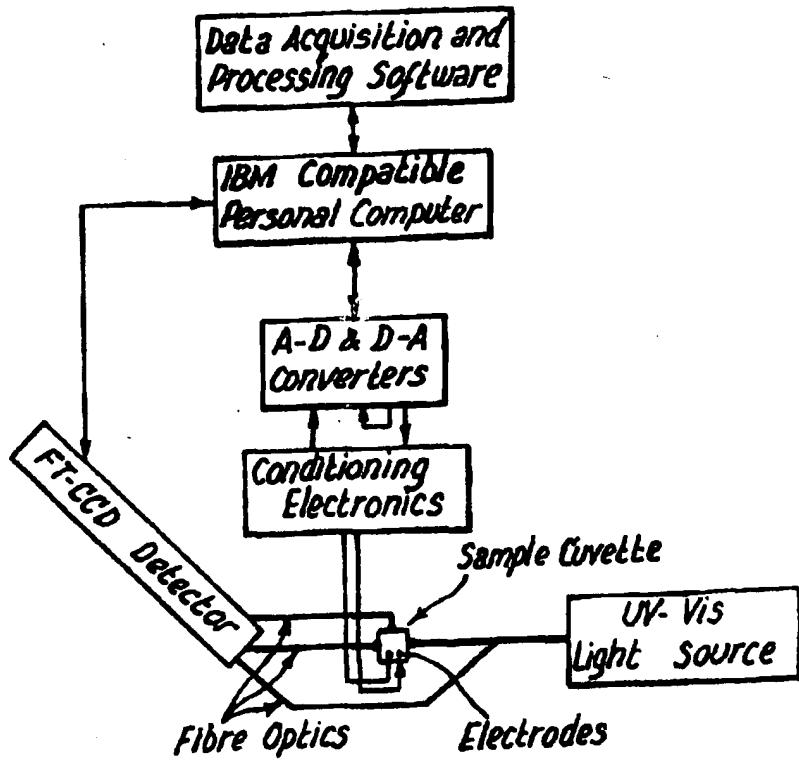


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<b>(54) Title:</b> DETECTOR FOR CHEMICAL ANALYSIS			
<b>(57) Abstract</b>			
<p>A tri-state detector which may be used for chemical analysis of flowing streams has a flow cell, light source, radiation detector and a micro-electrode chemical detector, the radiation detector being, for example, a Fourier Transform UV/visible spectrometer.</p>			
			

1        "Detector for Chemical Analysis"

2  
3        This invention relates to detectors and particularly  
4        tri-state detectors for use in chemical analysis of  
5        flowing streams. Such streams may be found in  
6        industrial processes, trade effluents, clinical  
7        solutions, or environmental waters, and will contain  
8        ions and molecules with a range of chemical properties.

9  
10      Process and environmental monitoring in the chemical  
11      industry is largely carried out in centralised  
12      laboratories which may be some distance away from where  
13      the sample is taken, resulting in a time delay between  
14      sampling and analysis. Should a fault develop in a  
15      manufacturing process, the time taken to detect the  
16      problem could be costly in terms of loss of production  
17      or damage to the environment. There is increasing  
18      need, therefore, in the chemical industry for the type  
19      of chemical detector that can be used on-line to give  
20      real-time chemical analysis of process or effluent  
21      streams, with the aim of improving product yield and  
22      quality whilst safeguarding the environment.

23  
24      Current provision for the chemical analysis of these  
25      sample streams is usually capable of dealing with only

1 one chemical property at a time. Most molecules may be  
2 characterised by their absorption or emission of light,  
3 or their electrical activity. At present, the methods  
4 of UV/visible absorbence, fluorescence and  
5 electrochemical analysis are all commonly used,  
6 separately, for the determination of chemical species  
7 (the analyte) in solution. Each method of detection is  
8 required depending on the type and concentration of  
9 analyte present in the sample. For example, not all  
10 molecular species can be detected by UV or fluorescence  
11 methods so that an electrochemical method may be more  
12 suitable for that particular analyte. In order to  
13 carry out a complete conventional analysis of complex  
14 sample may therefore require that the analysis is  
15 repeated with each of these three techniques in turn,  
16 resulting in long analysis times. Three different  
17 instruments will also be required, with obvious cost  
18 implications. The amount of chemical information that  
19 each technique can provide on its own is limited,  
20 whereas the combination of the three techniques will  
21 provide more detailed information regarding a  
22 particular sample.

23

24 According to the first aspect, the present invention  
25 provides a detector which is adapted to simultaneously  
26 investigate the following properties of a sample:

- 27 a) absorbance with respect to electromagnetic  
28 radiation;
- 29 b) fluorescence and/or emission; and
- 30 c) electrochemical properties.

31

32 The sample is preferably a flowing stream. It may be,  
33 for example, a process or effluent stream, or an eluent  
34 from a chromatography column.

35

36 Absorbance of the sample is preferably investigated

1       with respect to ultra-violet and/or visible radiation.

2

3       The detector may be capable of multi-wavelength  
4       measurement or analysis. It may be provided with one  
5       or more rapid scanning monochromators. This may allow  
6       rapid wavelength scanning when the detector operates in  
7       an absorbance and/or an emission mode; this facilitates  
8       rapid acquisition of chemical information relating to  
9       the sample. The detector may comprise one or more  
10      solid state monochromators.

11

12      The electrochemical properties or activity of the  
13      sample may be determined using an amperometric or  
14      coulometric technique. Preferably, pulsed amperometric  
15      detection (PAD), which can detect any sample molecule  
16      which has the ability to undergo an oxidation or  
17      reduction reaction at an electrode and produce an  
18      electrical current is used. A potential applied to the  
19      electrode is chosen to match the oxidation or reduction  
20      potential of an analyte species; the amount of current  
21      generated is proportional to the analyte concentration.

22

23      The detector may utilise a single microelectrode to  
24      conduct electrochemical analysis; this may have a  
25      surface diameter of less than 100 $\mu$ m. Alternatively, an  
26      array of several microelectrodes may be used, each of  
27      which may be set to a different measurement potential  
28      to allow selective detection and concentration  
29      measurement of each electroactive species present. The  
30      detector may incorporate a de-mountable electrode unit.  
31      This may comprise a microelectrode or microelectrode  
32      array for use in electrochemical analysis. The  
33      electrode unit may be removable to facilitate cleaning  
34      and restoration of a surface or surfaced of the  
35      electrode or electrodes; it may be disposable and this  
36      may alleviate problems of contamination of the

1     electrode surface or surfaces. A common problem with  
2     existing electrochemical detectors is that of  
3     contamination of the electrode surface leading to a  
4     loss in sensitivity.

5

6     According to a second aspect, the present invention  
7     provides a detector adapted to investigate  
8     electrochemical properties of a sample in which the  
9     detector comprises a removable electrode unit.

10

11    According to a third aspect, the present invention  
12    provides a detector adapted to investigate absorbance,  
13    fluorescence or emission of a sample in which the  
14    detector comprises one or more solid state  
15    monochromators.

16

17    According to a fourth aspect, the present invention  
18    provides a tri-state detector comprising a Fourier  
19    Transform UV/visible spectrometer.

20

21    An embodiment of the invention will now be described,  
22    by way of example only, with reference to the  
23    accompanying drawings of which:

24

25       Fig. 1 is a schematic perspective view of a  
26       detector;

27       Fig. 2 is a cross-section through Fig. 1; and  
28       Fig. 3 is a schematic diagram of a tri-state  
29       detector.

30    The detector which is illustrated schematically in the  
31    accompanying drawings has the ability to carry out a  
32    detailed chemical analysis of a flowing stream. The  
33    detector is compact, easy to use and relatively low-  
34    cost in comparison with single-component detectors  
35    presently available. It is a tri-state detector. It  
36    incorporates appropriate optical components and is

1       intended for use with appropriate data-handling  
2       softwar .

3

4       Referring to the drawings, light from a polychromatic  
5       light source 3 irradiates the face of an optically  
6       transparent flow cell 4. A monochromator 10, placed  
7       between the source 3 and the flow cell 4 is used to  
8       select radiation of specified wavelengths and the  
9       absorbance of the radiation by the analyte present in  
10      solution is measured in the x-plane by a  
11      photomultiplier 11. This incident radiation also  
12      serves as the excitation source for the emission mode.  
13      Simultaneously, fluorescent emission is measured at  
14      right angles to the incident radiation via a second  
15      monochromator 12, which is used for the wavelength  
16      selection of the emitted radiation, with a second  
17      photomultiplier 13. An electrochemical detector 15,  
18      which in this embodiment is a microelectrode or  
19      microelectrode array, is incorporated with a de-  
20      mountable electrode unit in the floor of the flow-cell  
21      so as not to interfere with the light path, and  
22      measures current as a function of electrode potential  
23      as electroactive analyte species are oxidised at the  
24      electrode. The electrochemical unit can be easily  
25      removed from the flow cell for cleaning/restoration of  
26      the electrode surfaces. A personal computer (not  
27      shown) and appropriate software is used for data  
28      collection and analysis. The flow cell may be  
29      interfaced to an existing separation system (eg HPLC),  
30      or may be incorporated within the pipework of an  
31      industrial plant or used for effluent monitoring.

32

33      Advantages of this detector include the following  
34      features:-

35

36      (1) Three detection modes incorporated within one

1           device allow simultaneous  
2           absorbance/emission/electrochemical measurement of  
3           a wide variety of both organic and inorganic  
4           species.

5

6       (2) Use of an electrochemical detector with a single  
7           microelectrode or an array of several  
8           microelectrodes, each set to a different  
9           measurement potential, allows selective detection  
10          and concentration measurement of each  
11          electroactive species present.

12

13      (3) In one embodiment, the detector makes use of two  
14           rapid scanning monochromators, preferably acousto  
15           optic tunable filters, for selecting the  
16           wavelength of light from the source and also that  
17           emitted as fluorescent radiation from the sample.  
18           Use of these devices allows rapid scanning across  
19           the chosen wavelength region of the  
20           electromagnetic spectrum, so that measurement of  
21           absorbance or fluorescent emission is not  
22           restricted to one wavelength at a time.

23

24      (4) The electro-chemical detector may be fabricated as  
25           a de-mountable unit. This allows easy access to  
26           the electrodes for cleaning and helps to alleviate  
27           the well-known problems of electrode  
28           contamination.

29

30      (5) The de-mountable electrode unit may be disposable.

31

32      (6) Shorter analysis times and increased sample  
33           throughput for liquid chromatography.

34

35      (7) Reduced cost compared to that of separate  
36           detectors.

1       (8) Capabl of *in situ* r al-time measurement of  
2           process or effluent streams.

3  
4       (9) Use of the specified monochromator allows rapid  
5           wavelength scanning and facilitates the rapid  
6           production of fluorescence/absorbance  
7           emission/excitation maps/ thus providing detailed  
8           chemical information regarding the analyte  
9           species.

10  
11      (10) Use of solid state monochromators and  
12           microelectrodes provides a compact device with no  
13           mechanical moving parts.

14  
15      (11) Use of state-of-the-art data acquisition software  
16           allows rapid data handling and storage.

17  
18      Referring to Fig. 3, light from a polychromatic source  
19           irradiates the face of the optically transparent flow  
20           cell via a fibre optic connection. This light passes  
21           through the sample solution where it is absorbed by  
22           optically active analyte molecules. The incident  
23           radiation also serves as the excitation source for the  
24           flourescence mode. The absorbance of radiation in the  
25           x-plane and the fluorescent emission in the y-plane are  
26           measured by the Fourier Transform (FT) spectrometer,  
27           via a network of fibre optic connections and switching  
28           devices.

29  
30      The electrochemical detector, which is a single  
31           microelectrode or microelectrode array, may be  
32           incorporated within a de-mountable unit in the body of  
33           the flow-cell so as not to interfere with the light  
34           path, or may comprise a unit immediat ly downstream of  
35           the optical flow cell. It is anticipated that th  
36           microelectrod unit will be easily removed from the

1 flow cell for cleaning or replacement, and can be made  
2 cheaply enough to be essentially disposable.

3

4 The entire system is controlled from a PC; the software  
5 for data acquisition and processing via an on-board A/D  
6 and D/A card has been developed using National  
7 Instruments Labview and LabWindows. The control and  
8 information flow between the main software package and  
9 the FT spectrometer control software is via the Dynamic  
10 Data Exchange (DDE) facility of Microsoft Windows.  
11 Experimental parameters such as number of scans  
12 acquired, output waveform to the electrochemical  
13 detector and presentation of data are managed via a  
14 graphical user interface in the Windows environment.

15

16 Through the use of specified optical components and  
17 appropriate data-handling software, the device will be  
18 compact, easy to use and relatively low-cost in  
19 comparison with equivalent combinations of single-  
20 component detectors presently available. The  
21 advantages of this tri-state detector may be summarised  
22 as follows:-

23

- 24 - Three detection modes incorporated within one  
25 compact device.
- 26 - Simultaneous measurement of light absorption and  
27 emission together with electrochemical activity of  
28 molecular species.
- 29 - Shorter analysis times and increased sample  
30 through-put for liquid chromatography.
- 31 - Reduced cost compared to that of separate  
32 detectors.
- 33 - Capable of in-situ real-time measurement of  
34 process or effluent streams.
- 35 - Measurement of absorption and fluorescence can be  
36 achieved in circa 1ms, allowing fast repetitive

1       scans in the situation where there is rapidly  
2       changing signal, (eg fast kinetic processes).  
3       - An easy-to-use Windows interface provides user-  
4       transparent data acquisition and processing.  
5       - The system can be used with existing liquid  
6       chromatographic equipment.  
7       - The detector components may also be used as stand  
8       alone instruments: UV/visible absorption  
9       spectrometer, fluorescence spectrometer and  
10      electrochemical detectors, thus offering a higher  
11      degree of flexibility in the use of all or any of  
12      the component parts of the detector.  
13      - Use of the FT Spectrometer, which is itself novel,  
14      ensures a compact device with no mechanical moving  
15      parts.  
16      - Easily de-mountable unit containing the  
17      microelectrode will simplify the task of electrode  
18      cleaning or replacement.  
19      - Information from the FT Spectrometer is available  
20      in digital form immediately with no hardware or  
21      software processing.  
22      - Potential exists for a portable version to be used  
23      with a lap-top PC in combination with a battery  
24      powered detection system.

25

26     The proposed tri-state detector is unique in the use of  
27     a Fourier Transform UV/visible spectrometer, to provide  
28     a compact device which has the capability of  
29     multiwavelength measurement. Also innovative is the  
30     design and fabrication of the de-mountable electrode  
31     unit, which may be disposable, to circumvent problems  
32     of contamination of the electrode surface.

33

34     Modifications and improvements may be incorporated  
35     without departing from the scope of the invention.

1       CLAIMS

2

3       1     A detector which is adapted to simultaneously  
4           investigate the following properties of a sample:  
5           a)   absorbance with respect to electromagnetic  
6           radiation;  
7           b)   fluorescence and/or emission; and  
8           c)   electrochemical properties.

9

10      2     A detector as claimed in Claim 1, wherein  
11           absorbance of the sample is preferably  
12           investigated with respect to ultra-violet and/or  
13           visible radiation.

14

15      3     A detector as claimed in Claim 1 or Claim 2  
16           capable of multi wavelength measurement or  
17           analysis.

18

19      4     A detector as claimed in any one of the preceding  
20           Claims provided with one or more rapid scanning  
21           monochromators.

22

23      5     A detector as claimed in Claim 4, wherein one or  
24           more of the monochromators are solid state  
25           monochromators.

26

27      6     A detector as claimed in any one of the preceding  
28           Claims adapted to use amperometric techniques for  
29           determining the electrochemical properties or  
30           activity of the sample.

31

32      7     A detector as claimed in any one of the preceding  
33           Claims, comprising a single microelectrode for  
34           conducting electrochemical analysis.

35

36      8     A detector as claimed in Claim 7 wherein th

1       single microelectrode has a surface diameter of  
2       less than 100 $\mu$ m.

3

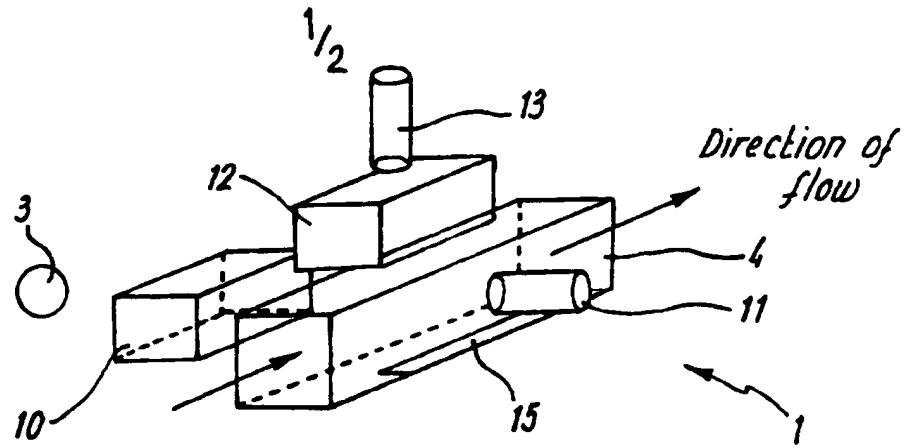
4       9      A detector as claimed in any one of Claims 1-7  
5       comprising an array of several microelectrodes,  
6       each being set to a different measurement  
7       potential to allow selective detection and  
8       concentration measurement of present electro-  
9       active species.

10

11      10     A detector adapted to investigate electrochemical  
12       properties of a sample in which the detector  
13       comprises a removable electrode unit.

14

15      11     A tri-state detector comprising a Fourier  
16       Transform UV/visible spectrometer.



*x - UV/ visible absorbance*

*y - fluorescence*

*z - direction of flow*

FIG. 1

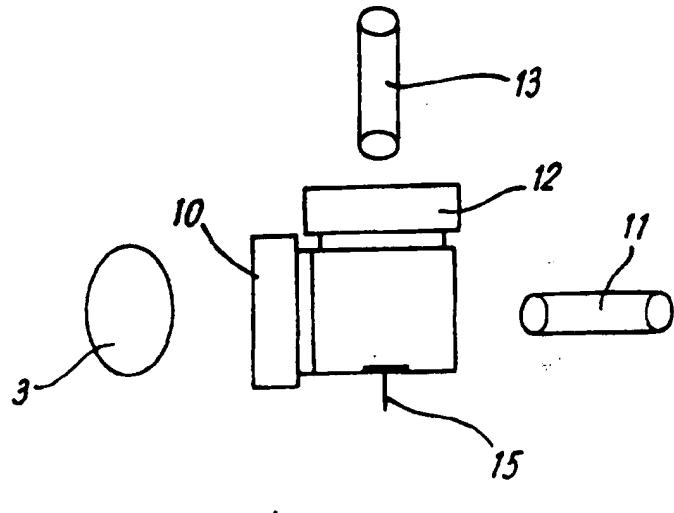


FIG. 2

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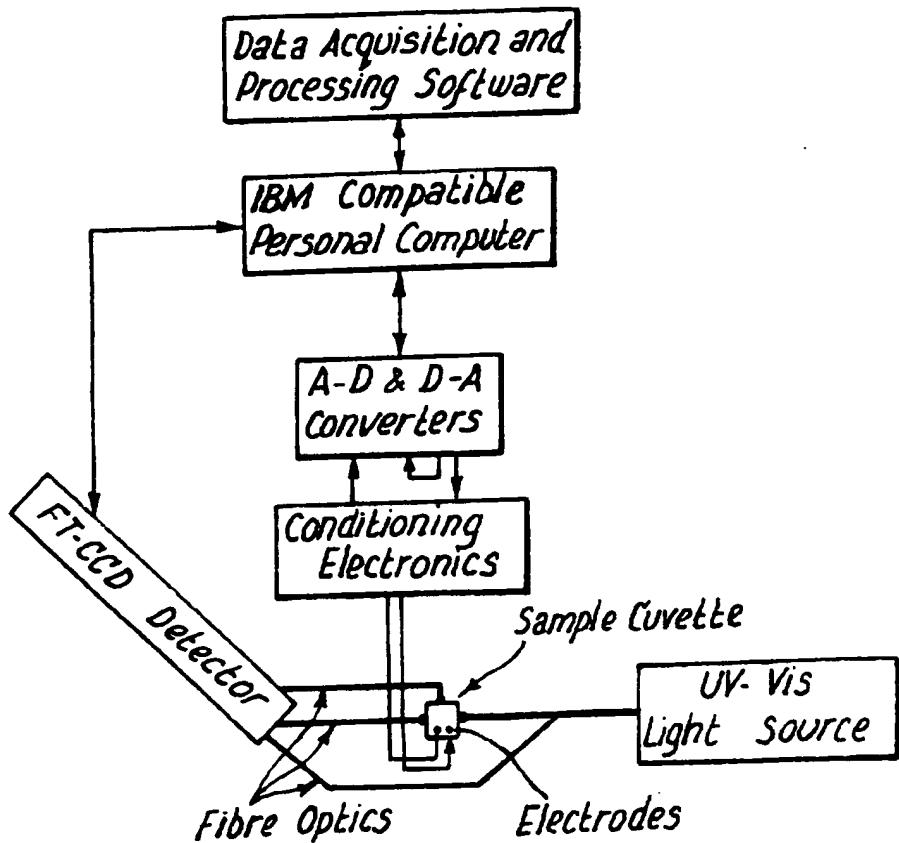


FIG. 3